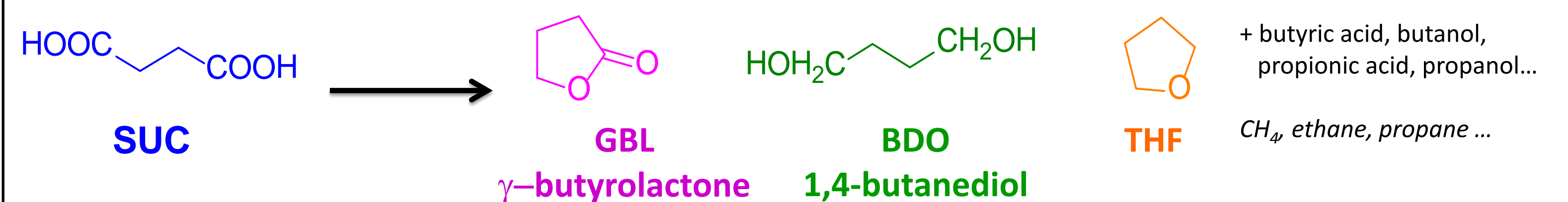


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OBJECTIVES:

- Evaluation of bimetallic catalysts **xwt%Re-2wt%Pd/TiO₂** in the aqueous phase catalytic hydrogenation of SUC
- Influence of preparation mode on activity of catalysts and selectivity to BDO



PREPARATION OF CATALYSTS

Monometallic catalyst
2wt%Pd/TiO₂

Support TiO₂ DT51 (S_{BET}=90 m²/g)
 Deposition-precipitation using K₂PdCl₄ and KOH
 Reduction by H₂ at 300°C-3h

Bimetallic catalyst **xwt%Re-2wt%Pd/TiO₂**
 Aqueous solution NH₄ReO₄
 using the monometallic catalyst

Successive impregnation (SI)

- Impregnation
- Evaporation and drying
- Activation by H₂ at 450°C-3h

3.4%Re-Pd-SI

Catalytic reduction (CR)

- Surface redox reaction: nH_{ads}-(Pd_s) + Re⁷⁺ → Re⁷⁻ⁿ-(Pd_s) + (7-n)H⁺
- Activation by H₂ at 450°C-3h

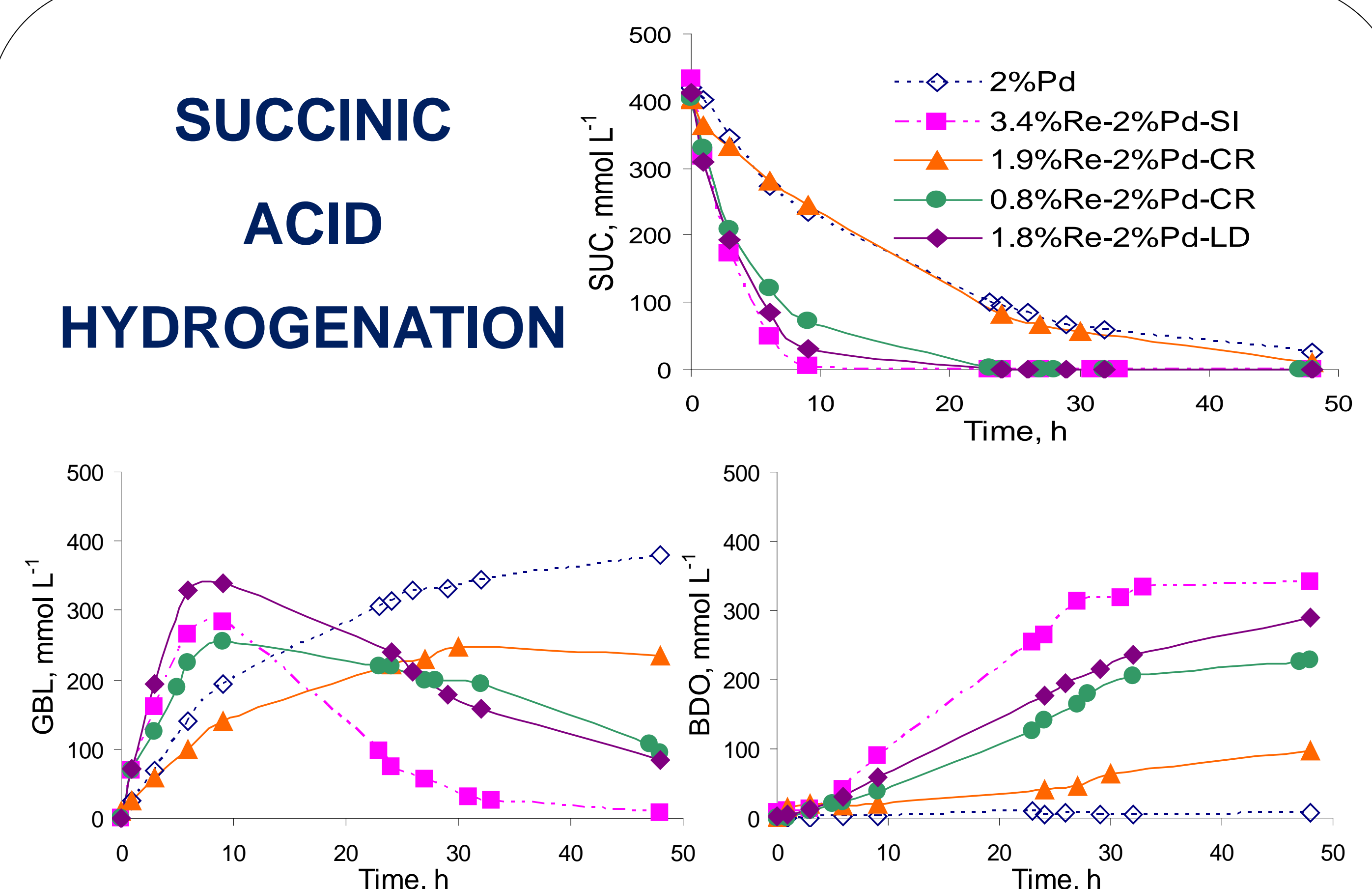
0.8 or 1.9 %Re-Pd-CR

In-situ treatment (liquid phase deposition)

- In the reactor, addition of precursor salt of Re.
- Activation by H₂ at 150 bar, 160°C, 1h.

1.8%Re-Pd-LD

SUCCINIC ACID HYDROGENATION



CR method (~1%Re) < LD (~2%Re) < SI (~4%Re)

CHARACTERIZATIONS

Dispersion

Cyclohexane dehydrogenation (CD)
= Structure insensitive (Re inactive)

Metal	H/Pd (%)	CD activity 270°C (mol.h ⁻¹ .g ⁻¹ .Pd)
2%Pd	27	2.9
0.8%Re-2%Pd-CR	10	2.2
1.9%Re-2%Pd-CR	8	2.0
3.4%Re-2%Pd-SI	21	2.4

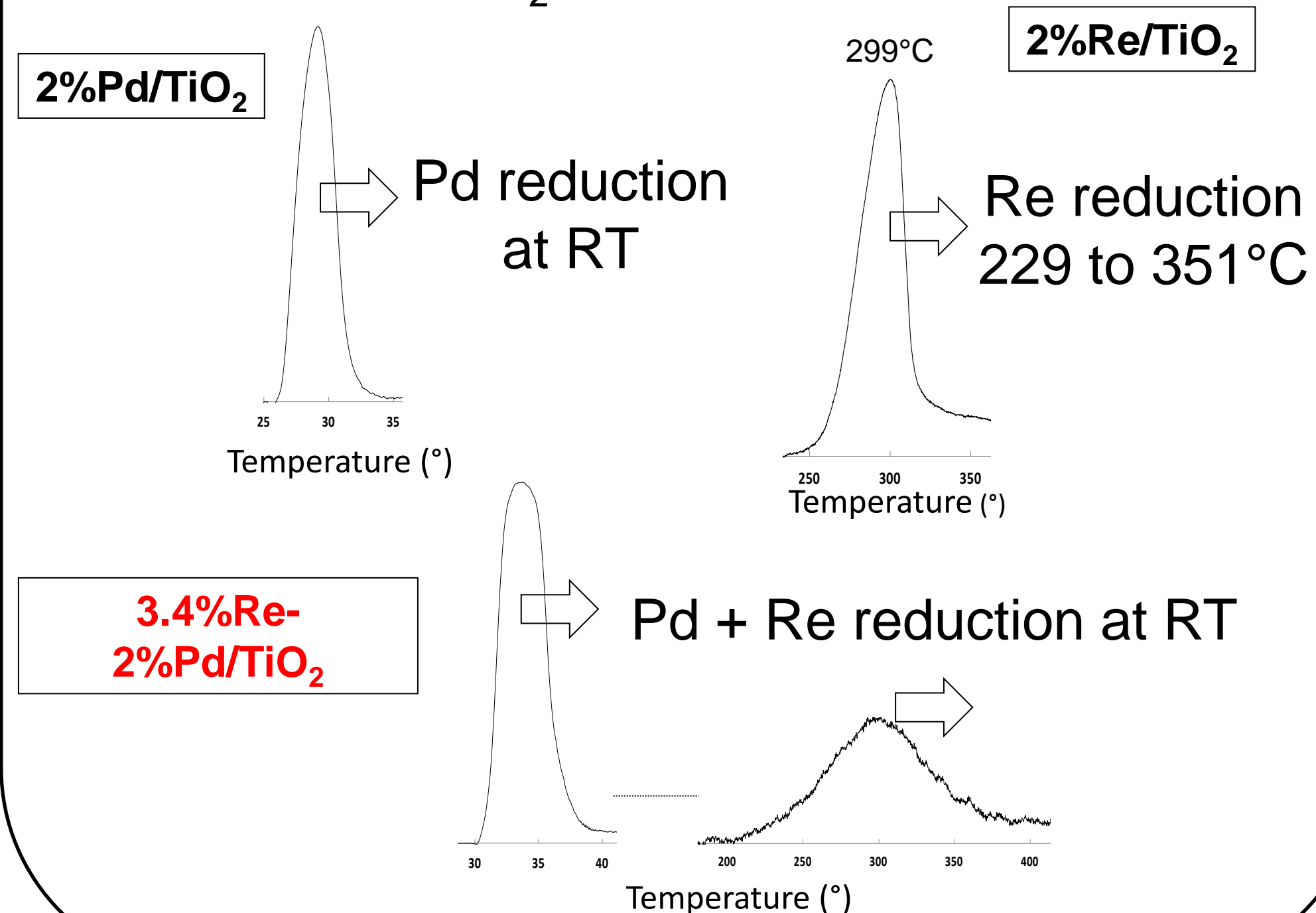
Addition of Re

- ↘ Pd accessibility
- ↘ Pd dehydrogenating activity

Re deposition onto Pd particles

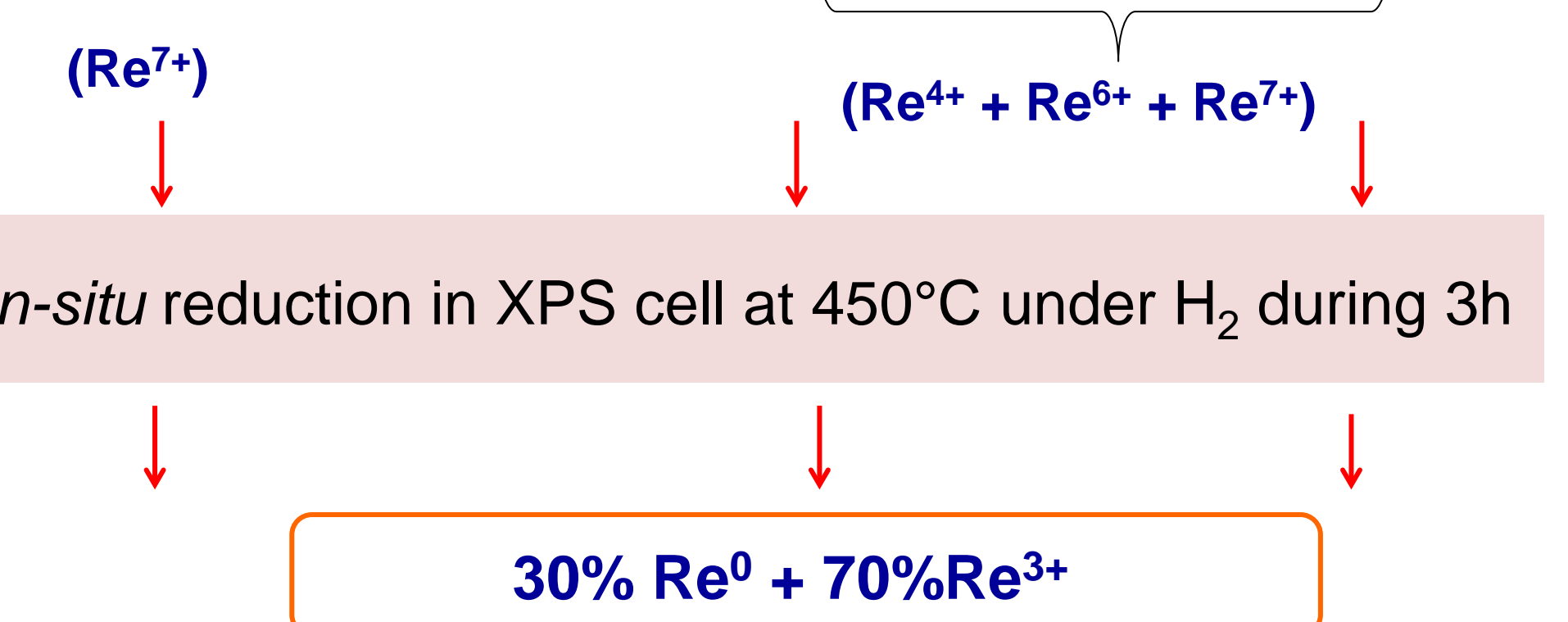
TPR

O₂ 300°C, N₂ at room temperature (RT),
TPR under 1%H₂/Ar from RT until 700°C



XPS

2.6%ReO₄NH₄-Pd-SI 2.6%Re-Pd-SI 2.6%Re-Pd-CR



- ✓ Re is readily oxidized in air.
- ✓ After reduction *ex-situ*, Re exists in different oxidation states.
- ✓ After reduction *in-situ* (H₂ at 450°C-3h), only ~30% is reduced up to Re⁰, the rest is Re³⁺

CONCLUSIONS

To obtain an optimum selectivity to BDO:
 the **SI** method required a high amount of Re (3-4wt.%),
 the **CR** and **LD** methods required lower loadings (0.6-0.9wt.% and 2wt.%, respectively).

**Best selectivity to BDO = 90%
at total conversion**

in the presence of 3.4%Re-2%Pd/TiO₂-SI

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