# CREAM

Chemical rections in macromolecules: a joint theoretical and experimental study

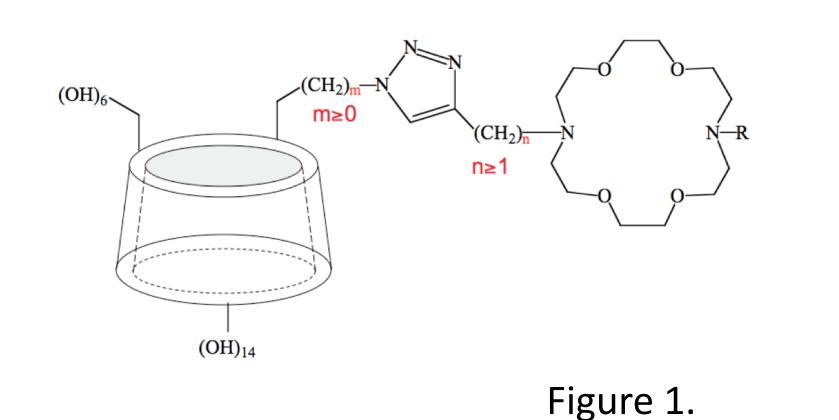
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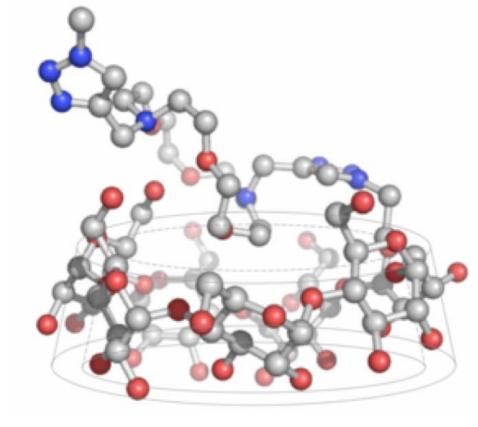


PI: Manuel F. Ruiz-Lopez, CNRS Research Director at SRSMC, University of Lorraine, Nancy. The team is composed of theoretical, physical and organic chemists from the same lab.

## Aims

- The CREAM project focuses on cyclodextrin (CD) applications as molecular reactors. The aims are: 1) development of more efficient theoretical tools to study chemical processes in macromolecules, 2) understanding environment effects in such reaction media and 3) synthesize of a new class of potential catalysts.
- Models of increasing complexity have been implemented and applied to theoretically investigate prototypical reaction mechanisms. Then a collaborative work between theoreticians and experimentalists has been carried out to synthesize and evaluate the catalytic properties of aza-crown  $\beta$ –CD derivatives (Figure 1) that are expected to selectively complex transition metals.







# Simulation, solvation effects in CD's

A large computational effort has been devoted to better understand how the properties of the guest (and then its reactivity) are modified under host-guest complex formation in aqueous solution. This has led us to: 1) evaluate the effective dielectric constant inside a  $\beta$ -CD, 2) analyze solvation effects at water/hydrophobic interfaces, 3) rationalize the observed effects on reaction rates (ester hydrolysis, Figure 2).

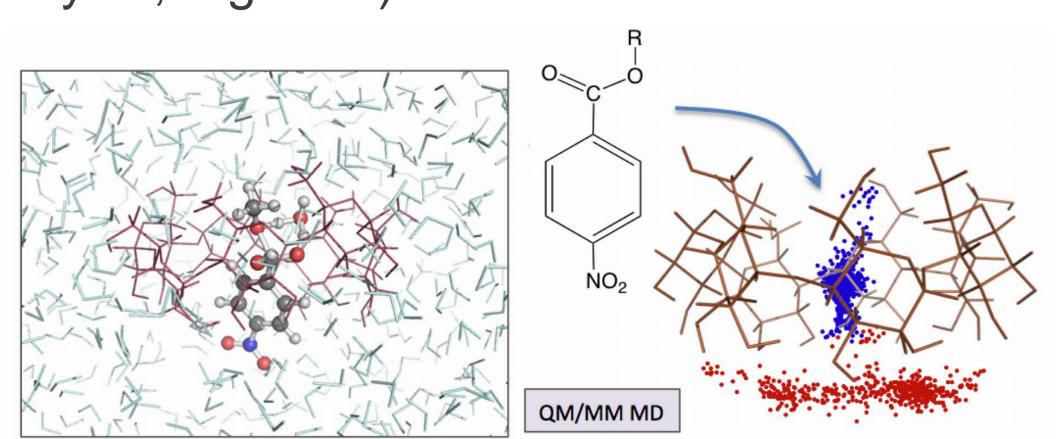
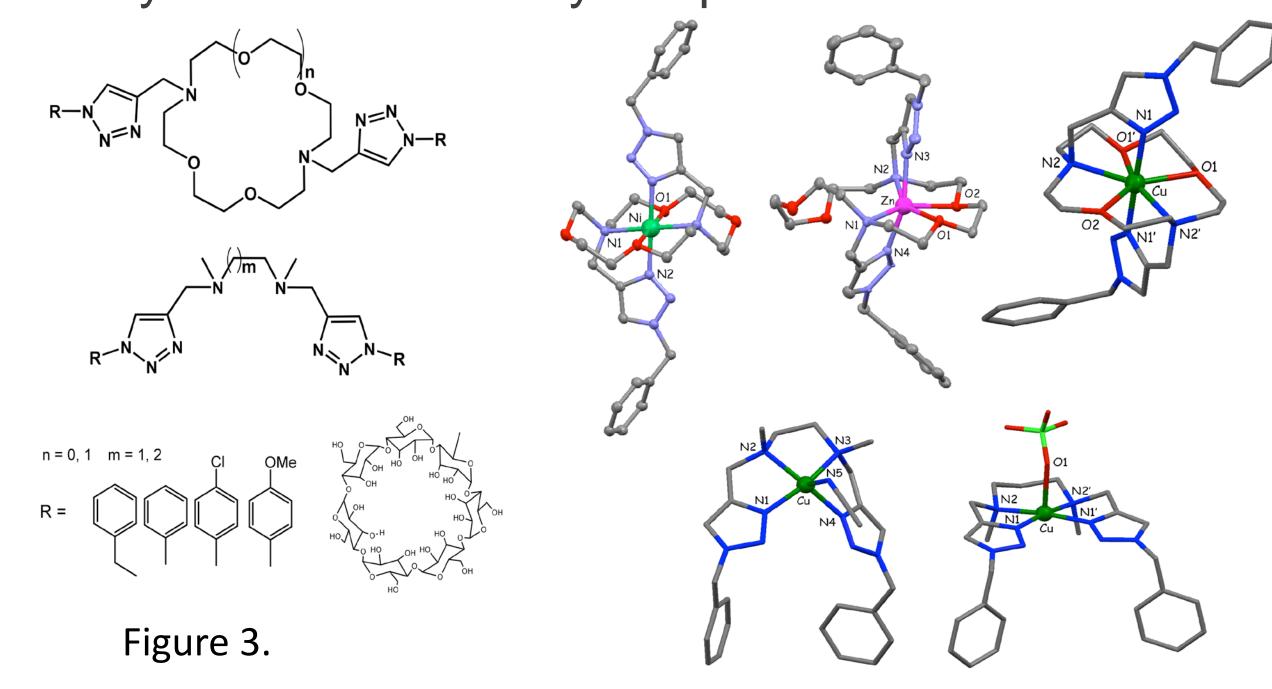


Figure 2.

## Synthesis, evaluation

The Huisgen Cu(I)-catalysed reaction has been chosen because of its high flexibility to synthesize cyclic and acyclic ligands. Then, 1:1 complexes with Cu<sup>2+</sup>, Ni<sup>2+</sup>, Zn<sup>2+</sup> have been investigated (Figure 3); the metal is axially or equatorially coordinated by the triazole groups and it displays penta-, hexa- or hepta-coordination. Cu<sup>2+</sup> complexes exhibit catalytic activity towards catechol oxidation, with acyclic systems being more efficient. This activity is not modified by the presence of the CD.



# From water to supercritical CO<sub>2</sub>

Supercritical CO<sub>2</sub> has attracted much attention as a green solvent. We have investigated the properties of CD's in this medium + their ability to form host-guest complexes (in collaboration with Prof. A. Marsura, SRSMC, and Dr. D. Barth, ENSIC-Nancy). The open/close dynamics of peracetylated CDs in scCO<sub>2</sub> has been described through Molecular Dynamics simulations (Figure 4). The results encourage us to continue further work in this direction.

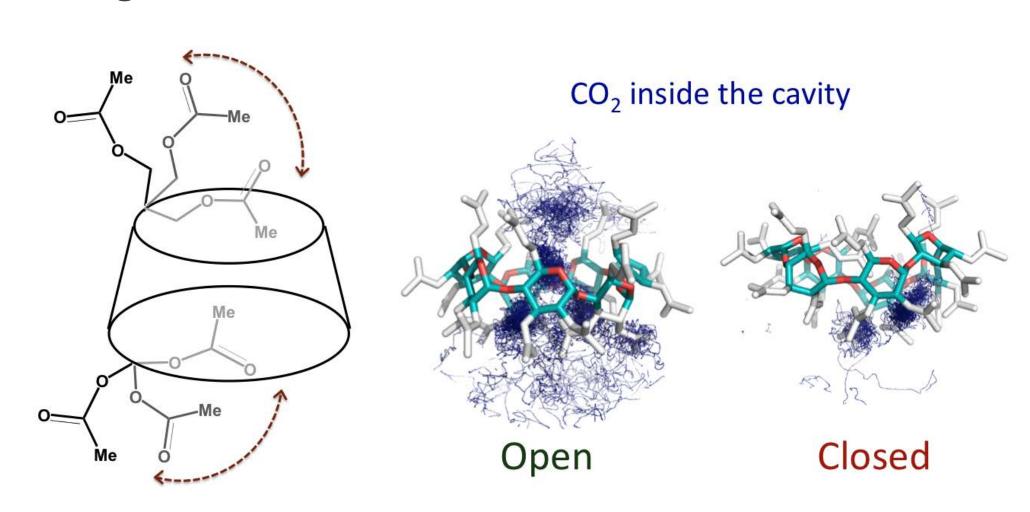


Figure 4.

### CONTACT:











