# CAPROCELL

Pure cellulose production from wood by an environmentally friendly process using catalysed hydrogen peroxide

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### Background and objectives

Cellulosic pulps for paper production are obtained after wood delignification (cooking) and bleaching with chlorinated agents to remove residual lignin. The use of these cellulosic pulps in chemical applications (cellulosic textiles and plastics) is not possible because of the cellulose contamination with hemicelluloses and lack of reactivity. Special delignification (cooking) processes have to be used which are more complicated and which do not allow to produce cellulosic pulps for both paper and cellulose applications. This project aims at developing a new environmentally friendly catalytic bleaching process, using hydrogen peroxide, for pure cellulose production from a wood pulp of paper quality obtained with conventional cooking process. Cu(II)-diimine complexes are used as catalyst for  $H_2O_2$ delignification.

#### Screening of Cu(II)-di-imine catalyst

Quantification of the veratry alcohol consumption (non phenolic lignin model) and veratryl aldehyde production after the catalyzed oxidation.



Fig 3. Oxidation of the veratryl alcohol into aldehyde and acid, in the presence of the Cu(II)-phenanthroline complex

Table 1. Selected di-imine ligands (22 ligands from several families were tested, with or without substituents R<sub>1</sub>, R<sub>2</sub>, R<sub>3</sub>, R<sub>4</sub>)



Cu(II)-phenanthroline type complexes are the most active.

Produced celluloses and technical and economical evaluation of the process

## Catalyzed $H_2O_2$ delignification by the Cu(II)-phenanthroline complex

Mechanism explaining the higher lignin depolymerization with the catalyst (study using Lignin Model Dimer, i.e. LMD):



**1st step:** oxidation of the alcohol group in the C- $\alpha$  position catalyzed by the Cu(II)-phenanthroline complex in alkaline medium - (no reaction without the catalyst). LMD1 is oxidized into LDM5

Fig 1. LMD1 = non phenolic Lignin Model Dimer 1-(3,4-diméthoxyphényl)-2-(2-méthoxy-phénoxy) propane-1-ol



**2nd step:** cleavage the  $\beta$ -O-4 ether OŤ linkages of LMD5 in alkaline medium (does not require the catalyst)

A bleaching/purification process has been developed which includes  $H_2O_2$  delignification, catalyzed (or not), and pre- and post- chlorine free treatments (cold caustic extraction E and ozonation Z)

Table 2. Celluloses properties

| Cellulose                  | Brightness,<br>%ISO | Viscosity,<br>mL/g | S18,<br>% | Cu,<br>ppm | Heavy metals, ppm |
|----------------------------|---------------------|--------------------|-----------|------------|-------------------|
| Viscose target             | 90-92               | 400-600            | 3-3.5     | 1          | <10               |
| (aQ)(OP)EZ                 | 90.3                | 874                | 3.3       | <5         | <0.5              |
| (aQ)(OP) <sub>cat</sub> EZ | 91.2                | 437                | 3.4       | <5         | <0.5              |

#### The produced celluloses exhibit similar properties as **viscose** pulps in terms of purity.

Table 3. Economical evaluation of the integration of the process in an existing bleached pulp production line (10% of the total production is used for pure cellulose production)

| € per produced tones | Cellulosic pulp | Pure cellulose    |                  |  |
|----------------------|-----------------|-------------------|------------------|--|
| Targeted sale price  | 580             | 880               |                  |  |
|                      |                 | Catalyzed process | Without catalyst |  |
| Wood cost            | 350             | 432               | 432              |  |
| Chemicals cost 50    |                 | 173               | 53               |  |
| Gross profit         | 180             | 275               | 395              |  |



The gross profit is better for pure cellulose than for cellulosic pulp. The process is working without catalyst, but the cellulose viscosity is too high for viscose application, (acceptable for other uses).

Fig 2.  $\beta$ -O-4 ether linkages in alkaline medium (LMD=Lignin Model Dimer)

